

Mechanics:Physical ScienceNewton's laws of motion. $s = \frac{1}{2}a \cdot t^2 + a \cdot t + s_0$ 1. Inertia $v = a \cdot t + v_0$ 2. $F_{net} = m \cdot a$ a = acceleration3. action = -reactiona = acceleration

speed is scalar, magnitude only vs. velocity is a vector, both magnitude and direction

Work and Energy	Gravity
Kinetic: $KE = \frac{1}{2} m \bullet v^2$	$\mathbf{s} = \frac{1}{2}\mathbf{g} \bullet \mathbf{t}^2 + \mathbf{g} \bullet \mathbf{t} + \mathbf{s}_0$
Potential : $PE = m \bullet g \bullet y$	$\mathbf{v} = \mathbf{g} \bullet \mathbf{t} + \mathbf{v}_0$
Work = $F \bullet s = E_2 - E_1$	$a=g \cong -9.8 \text{ m/s}^2 \cong -32 \text{ ft/s}^2$
Power = work/time	$g = G \bullet m_1 \bullet m_2/d^2$
	$G = 6.67 \times 10^{-11} \text{ N} \cdot \text{m/kg}^2$

Chemical bonds

Covalent: electron sharing, strong bonds each atom is attracted for shared electrons. Ionic: electron gaining/losing, change in charge attracts atoms Metal: electrons loosely bonded to metal nuclei. Sea of electrons permits heat, electricity and shaping of metal.

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Mechanics: Newton's laws of motion.

1. Inertia

2. $F_{net} = m \bullet a$

3. action = -reaction

Physical Science Kinematic Equations $s = \frac{1}{2}a \bullet t^2 + a \bullet t + s_0$ $v = a \bullet t + v_0$ a = acceleration

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Physical Science

Electricity and Magnetism V, voltage (volt, V) = PE/charge or I•R I, current (ampere, A) = charge/time or voltage/resistance R, resistance (ohm, Ω) = voltage/current P, power (watt, W) = I•V q, charge (coulomb, C) = 1C \cong charge on 6.25x10¹⁸ e⁻ Coulombs Law: F = k•q1•q2/d² Faraday's Law: voltage induced \cong number of loops•B field/time F = q•v•B V, velocity (meter/second) B, magnetic field strength (tesla, T)

Atom, Nuclear, and Chemistry

Mass number, the number of nucleons $p^+ + n^0$. Atomic number, the number of protons p^+ . Number of neutrons = mass # - atomic #. Isotope is an atom with differing # of neutrons n^0 , has radioactive properties. Energy is released when mass is converted into E according to $E = mc^2$.

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